Transforming our understanding of methanation: One-step CO₂-conversion in Microbial Electrolysis Cells

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Introduction

Power-to-Methane with biological catalysis is typically comprised of two steps – electrolysis and methanation – to convert electricity, water and carbon dioxide into renewable methane (Figure 1). Based on the rapid technology scale-up over the last decade (Figure 2) Electrochaea provides a superior solution for energy storage and for the future provision of climate neutral fuel, contributing to the mandatory reduction of the CO₂footprint.



Rapid Technology De-risking and Scale-up

Commercial-Scale Field Trial Preparing for market entry with a commercial-scale demonstration unit, using an optimized reactor, Copenhagen, Denmark

Pre-Commercial Field Trial

Process demonstration in a 5m³ stirred tank bioreactor using raw biogas, Foulum, Denmark

Lab-Scale Field Trial

Biocatalytic capability test with raw biogas

CO₂ from biogas or waste water treatment plants, geothermal gases or fermentation off gases etc.

Figure 2: Electrochaea's technology scale-up from basic research to 1 MW field trial. industrial-scale PtG plants are located in Denmark, Switzerland and the US (NREL, CO). The European units are injecting biomethane in the natural gas grids. The gas is also compliant with SoCalGas Rule 30.



1 MV

50 kW

Basic Research In Dr. Mets' laboratory at the University of Chicago



Power-to-methane in one reactor

Electrochaea has also been developing a technology where *electrolysis* and *bio-methanation* are merged into one single reactor, a Microbial Electrolysis Cell (MEC)¹. This transition from a 2-step to a 1-step process enables the adaptation of Electrochaea's bio-methanation technology into smaller gas producer facilities and lower electricity availabilities. The benefits of MEC as a platform reactor for biological CO_2 conversion to CH_4 (electro-biomethanogenesis) include:

- ✓ Elimination of the capital cost associated with having a separate electrolyzer and stirred bioreactor
- \checkmark H₂ production in the same reactor as the microorganism converting CO₂ into CH₄, thus overcoming productivity limitations associated with low solubility and poor mass transfer of H₂ in water



MEC fundamentals

- \checkmark MEC uses renewable electricity as the energy source for the production of methane from CO₂
- ✓ The system consists of two electrodes, anode and cathode, separated by a membrane
- \checkmark At the anode, the water oxidation reaction takes place yielding 2 H⁺ and 8 electrons
- Electrons flow through an external electrical circuit to the cathode, whereas protons migrate through the membrane to the cathode to maintain electroneutrality
- \checkmark At the cathode, the protons and electrons are used to produce methane
- \checkmark The reaction at the cathode is catalyzed by electrochemically active microorganisms.



Figure 5: Bioelectrochemical performance of methane production in a Microbial Electrolysis Cell from CO₂ and electricity. Electrochaea's proprietary biocatalyst is a Methanothermobacter thermautotrophicus strain² developed in in Prof. Mets Lab (University of Chicago, IL). The density of the culture placed at the biocathde duplicates during the course of the experiment.

Figure 6: Relative methane production in % reported in peer reviewed articles published between 2009 and 2020³. The orange bar represents the relative methane production (%) achieved with the 1-step process at Electrochaea.

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conversion of carbon dioxide (CO2) to low-carbon methane by bioelectromethanogenesis process in

microbial electrolysis cells: the current status and future perspective. Bioresource technology.

sized enterprises (SMEs)



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